

B.Sc. DEGREE (C.B.C.S.S.) EXAMINATION, MARCH 2015**Sixth Semester****Core Course—NUCLEAR AND PARTICLE PHYSICS**

[Common for Model I B.Sc. Physics, Model II B.Sc. Physics, B.Sc. Physics,
B.Sc. Physics-EEM and B.Sc. Physics-Instrumentation]

Time : Three Hours

Maximum Weight : 25

Part A (Objective Type)*Answer all questions.**Each bunch of four questions carries a weight of 1.***Bunch I**

1. Isomers are the atoms having :

- | | |
|--------------------------|-------------------------|
| (a) Same mass number. | (b) Same atomic number. |
| (c) Same neutron number. | (d) Both (a) and (b). |

2. Which of the following nucleus is more stable ?

- | | |
|---------------------|--------------------|
| (a) Oxygen nucleus. | (b) Xenon nucleus. |
| (c) Iron nucleus. | (d) Lead nucleus. |

3. Nuclear force is :

- | | |
|------------------------|--------------------------|
| (a) Long range force. | (b) Electrostatic force. |
| (c) Short range force. | (d) All of the above. |

4. Which of the following decay atomic number remains the same ?

- | | |
|------------------|-----------------------|
| (a) Beta decay. | (b) Gamma decay. |
| (c) Alpha decay. | (d) Both (a) and (b). |

Bunch II

5. The nuclear shell model is also named as :

- | | |
|---------------------|---------------------------------|
| (a) Electron model. | (b) Independent particle model. |
| (c) Atom model. | (d) None. |

6. Packing fraction is associated with nucleus in terms of :

- | | |
|--------------------|-----------------------|
| (a) Stability. | (b) Mass defect. |
| (c) Nuclear force. | (d) All of the above. |

Turn over

7. Ionisation chamber is filled with which of the following gas :

- (a) Methyl bromide. (b) Hydrogen.
(c) Nitrogen. (d) None.

8. Gamow's theory of alpha decay depends on :

- (a) Classical physics. (b) Quantum physics.
(c) Both (a) and (b). (d) None.

Bunch III

9. The charge of a neutrino is :

- (a) Positive. (b) Negative.
(c) Neutral. (d) Either positive or negative.

10. Internal conversion is a :

- (a) Single step process. (b) Double step process.
(c) Either single or double. (d) Both (a) and (b).

11. In a nuclear reaction which is conserved :

- (a) Charge. (b) Nucleons.
(c) Parity. (d) All of the above.

12. Nuclear fission in a nuclear reactor is :

- (a) Chain reaction. (b) Controlled chain reaction.
(c) Uncontrolled chain reaction. (d) All of the above.

Bunch IV

13. In nuclear reactors ——— neutrons are used.

- (a) Slow. (b) Fast.
(c) Very fast. (d) Super fast.

14. Which one of the following is a baryon ?

- (a) Photon. (b) Electron.
(c) Sigma. (d) Kaon.

15. A meson have how many quarks :

- (a) One. (b) Two.
(c) Three. (d) None.

16. The spin value of a baryon is :

- (a) Zero. (b) $1/2$.
(c) $-1/2$. (d) None.

Part B (Short Answer Questions)

Answer any five questions.

Each question carries a weight of 1.

17. Define mass defect.
18. Write short note on nuclear density.
19. Give few names of nuclear detectors.
20. Define mean life of a radioactive element.
21. State Geiger-Nuttall law.
22. Give one example of artificial radio activity.
23. What is a thermonuclear reaction ?
24. Give the quark model for a proton.

(5 × 1 = 5)

Part C (Short Essay/Problems)

Answer any four questions.

Each question carries a weight of 2.

25. Give the meson theory of nuclear force.
26. Find the energy release, if two, H^2 nuclei can fuse together to form ^4He nucleus. The binding energy per nucleon of H^2 and He^4 is 1.1 MeV and 7.0 MeV respectively.
27. Write a note on proton-proton cycle.
28. Calculate the Q-value of the following nuclear reaction in MeV :—
 $^{14}\text{N} + 2^4\text{He} \rightarrow ^{17}\text{O} + ^1\text{H} + Q$
Where, $N^{14} = 14.003074$, $\text{He}^4 = 4.002604$, $\text{O}^{17} = 16.99913$, $\text{H}^1 = 1.007825$.
29. Write a note on primary and secondary cosmic rays.
30. Describe about radiation hazards and their disposal.

(4 × 2 = 8)

Part D (Essay)

Answer any two questions.

Each question carries a weight of 4.

31. Describe the working function of a Bainbridge's mass spectrometer.
32. Give the Gamow's theory of alpha decay and explain it in detail.
33. Discuss the various types of elementary particles in detail.

(2 × 4 = 8)